

Carbostannylation of Alkynes Catalyzed by an Iminophosphine–Palladium Complex

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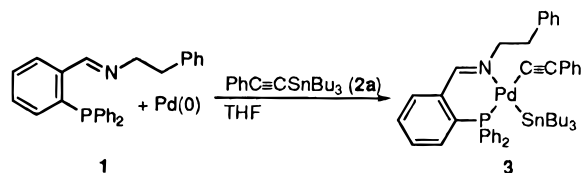
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Carbometalation of alkynes generates cis-substituted alkenylmetals and is one of the most useful reactions for stereoselective olefin synthesis, since the resulting alkenylmetals can be transformed further to variously substituted ethylenes.¹ In particular, carbocupration,² zirconium-catalyzed carboalumination³ and nickel-catalyzed carbocation⁴ have high synthetic potential due to wide applicability. Although alkenylstannanes are useful synthetic precursors for various olefinic targets,⁵ no report has been published on the transition-metal-catalyzed carbostannylation of alkynes.^{6–8} Here, we report the palladium-catalyzed carbostannylation of alkynes using alkynylstannanes.

We have already reported that tributyl(phenylethynyl)tin (**2a**) adds oxidatively to a palladium(0) complex coordinated by *N*-(2-(diphenylphosphino)benzylidene)-2-phenylethylamine (**1**) and that the resulting oxidative adduct (**3**) is involved in the catalytic cycle of the palladium-catalyzed coupling of **2a** with aryl iodides (Scheme 1).⁹ We envisaged that palladium complex **3** should react with alkynes to give carbostannylation products. This turned out to be the case.

Treatment of **2a** with a 1:2 mixture of [PdCl(π -C₃H₅)]₂-**1** (5 mol % of Pd) under an acetylene atmosphere (1 atm) in THF at 50 °C for 2 h gave tributyl(*Z*)-2-(phenylethynyl)ethenyltin (**5a**)¹⁰ in 81% yield¹¹ as a single isomer through an exclusive syn-addition (Scheme 2). The use of triphenylphosphine (2 equiv to palladium) in place of **1** gave only 48% yield of **5a** in a prolonged period (43 h). A Pd(0)–1,3-bis(diphenylphosphino)propane

Scheme 1



Scheme 2

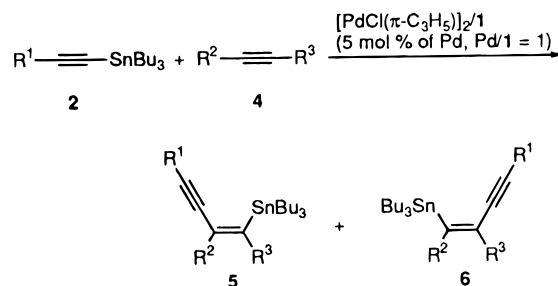


Table 1. Carbostannylation of Alkynes Catalyzed by Iminophosphine (**1**)–Palladium^a

entry	R ¹	R ²	R ³	temp (°C)	time (h)	yield (%) ^b	prod(s)	5/6 ^c
1	Ph (2a)	H	H (4a) ^d	50	2	81	5a	
2	Ph (2a)	CO ₂ Et	H (4b)	50	3	78	5b, 6b	20/80
3	Ph (2a)	CO ₂ Et	Me (4c) ^e	90	90	57	5c, 6c	1/>99
4	Ph (2a)	Ac	H (4d)	50	4	76	5d, 6d	15/85
5	Ph (2a)	Ph	H (4e)	50	21	81	5e, 6e	92/8
6	Ph (2a)	4-CH ₃ C ₆ H ₄	H (4f)	50	44	82	5f, 6f	91/9
7	Ph (2a)	EtO	H (4g) ^f	50	5	52	5g, 6g	>99/1
8	Bu (2b)	H	H (4a) ^d	50	4	66	5h	
9	Bu (2b)	CO ₂ Et	H (4b)	50	16	72	5i, 6i	12/88
10	Bu (2b)	Ph	H (4e)	50	29	80	5j, 6j	92/8

^a The reaction was carried out in THF (5 mL) using an alkynylstannane (0.459 mmol) and an alkyne (1.38 mmol) in the presence of iminophosphine **1** (0.022 mmol) and [PdCl(π -C₃H₅)]₂ (0.011 mmol).

^b Isolated yield based on the alkynylstannane is given. ^c Determined by ¹H or ¹¹⁹Sn NMR. ^d The reaction was carried out under an acetylene atmosphere (1 atm). ^e Solvent = dioxane. ^f Ethoxyacetylene (0.459 mmol) was used.

complex, which also was shown to be added oxidatively by **2a**,⁹ was much less effective to give **5a** in 28% yield even after 22 h.

The carbostannylation of various alkynes catalyzed by the Pd-**1** catalyst was next examined (Scheme 2, Table 1). The reaction of **2a** with ethyl propiolate (**4b**) gave carbostannylation products consisting of regioisomers **5b** and **6b** in a 20/80 ratio (entry 2). The carbostannylation of ethyl 2-butyrate (**4c**) resulted in higher regioselectivity, though higher temperature and longer reaction time were required (entry 3). A ketonic acetylene, 1-butyne-3-one (**4d**), reacted with **2a** smoothly with a regioselectivity similar to **4b** (entry 4). The reaction of arylacetylenes **4e** and **4f** with **2a** was relatively slow to give the corresponding carbostannylation products in high yields (entries 5 and 6) but with a reversed regioselectivity. The reaction of ethoxyacetylene (**4g**) also proceeded with high regioselectivity with the preference for **5g** over **6g** (entry 7). Tributyl(1-hexyn-1-yl)tin (**2b**) also reacted with alkyne **4a**, **4b**, or **4e**, giving the corresponding alkynylstannanes with the regioselectivity similar to **2a** (entries 8–10).¹²

(12) No isomer other than **5** and **6** was obtained out of the four possible isomers in use of alkynes **4b–g**. The hydrolysis of the alkynylstannanes to the corresponding alkenes revealed that the stannyl group in **5** had a proton at the geminal position, whereas that in **6** had an R² group. Syn-addition in use of acetylene led us to the conclusion that both **5** and **6** are also syn.

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(6) Carbostannylation of alkynes using a particular combination of substrates, (stannylethynyl)amines and dimethyl acetylenedicarboxylate, is reported. Himbert, G. *J. Chem. Res. S* 1979, 88–89.

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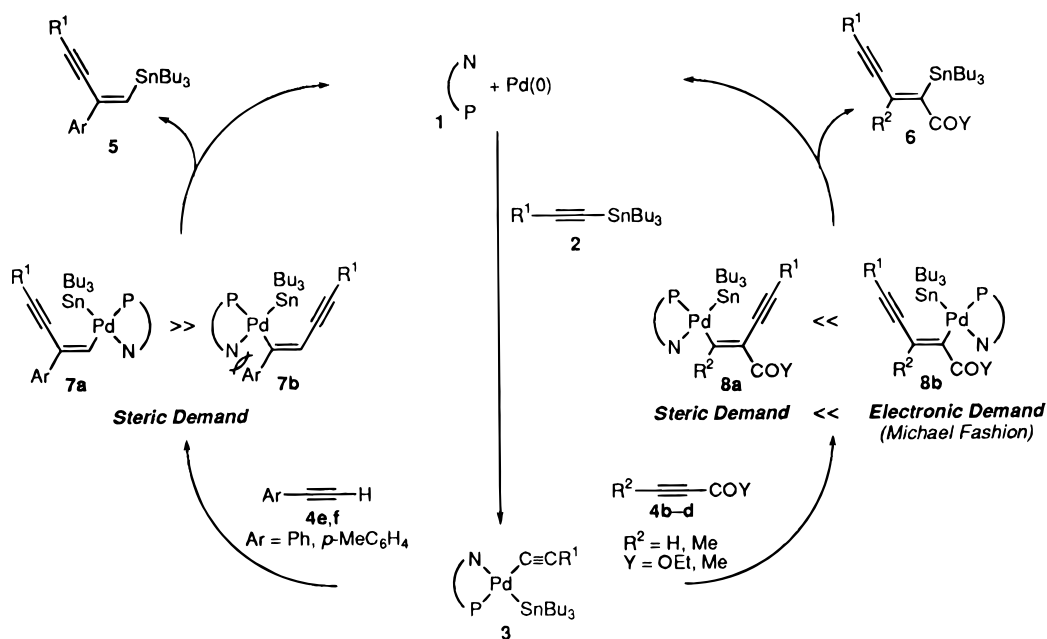
(8) Carbostannylation may alternatively be accomplished by carbocupration of alkynes followed by quenching with tin halides/triflate. Westmijze, H.; Meijer, J.; Vermeer, P. *Recl. Trav. Chim. Pays-Bas* 1977, 96, 194–196.

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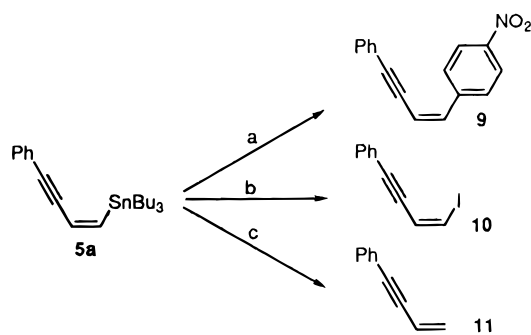
(10) Configuration of alkynylstannane **5a** could not be determined directly, because the olefinic protons of **5a** had the same chemical shift in ¹H NMR. The (*Z*)-configuration of **5a** was confirmed after the transformation to the corresponding alkenyl iodide (**10**) by iodolysis (Scheme 4). The coupling constant between olefinic protons of **10** was 8.3 Hz, which is typical to a *cis*-disubstituted ethylene. Iodolysis of α - or β -(alkynyl)alkenylstannanes to the corresponding (alkynyl)alkenyl iodides proceeds with retention of configuration: Stracker, E. C.; Zweifel, G. *Tetrahedron Lett.* 1991, 32, 3329–3332.

(11) The yield based on alkynylstannane should be lower than 95%, because 5% of alkynylstannane would be consumed for the reduction of Pd(II) to Pd(0).

Scheme 3



Scheme 4



(a) 4-O₂NC₆H₄I (0.92 equiv), Pd₂(dba)₃, (2-furyl)₃P (5 mol % of Pd, Pd/(2-furyl)₃P = 1/4), toluene, 90 °C, 13 h, 85% (based on 4-O₂NC₆H₄I). (b) I₂ (1.4 equiv), THF, 0 °C, 40 min, 91%. (c) conc. HCl, THF, rt, 1 h, 82%.

The catalytic cycle should first involve the oxidative addition of an alkylnylstannane to the Pd(0) complex as discussed before.⁹ Successive insertion of an alkyne to the C–Pd bond (carbopalladation)¹³ of **3** followed by reductive elimination is likely to afford the carbostannylation product and regenerate the Pd(0) complex (Scheme 3). The opposite regioselectivity observed in use of arylacetylenes and alkynes having a carbonyl group can

(13) Some reaction may take place through the stannylpalladation of an alkyne. A catalytic cycle of the palladium-catalyzed hydrostannylation–cyclization of 1,6-diynes is discussed to proceed in similar dual pathways. Lautens, M.; Smith, N. D.; Ostrovsky, D. *J. Org. Chem.* **1997**, *62*, 8970–8971.

be explained by the followings. In the cases of arylacetylenes, carbopalladation giving alkenylpalladiums would prefer **7a** to **7b** by steric reason. Accordingly, alkenylstannanes **5** are afforded as main products. In contrast, electron-deficient alkynes **4b**, **4c**, and **4d** are likely to suffer the addition of the alkylnyl group in a Michael fashion giving **6** predominantly through **8b**.

Finally, we confirmed the utility of the carbostannylation products by the transformation of carbostannylation product **5a** to various compounds (Scheme 4). The Pd–tri(2-furyl)phosphine-catalyzed coupling¹⁴ of **5a** with 4-nitroiodobenzene gave enyne **9** in 85% yield.¹⁵ Iodolysis or hydrolysis of **5a** afforded the corresponding alkenyl iodide (**10**) or enyne (**11**) in good yield, respectively.

In conclusion, we have demonstrated that the carbostannylation of alkynes takes place with alkylnylstannanes to give conjugated (stannyl)enyne in a stereoselective manner. Studies on the details of the mechanism as well as synthetic applications to various unsaturated substrates and organostannanes are in progress in our laboratories.

Supporting Information Available: Detailed experimental procedures including spectroscopic and analytical data (5 pages, print/PDF). See any current masthead page for ordering information and Web access instructions.

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(15) A [PdCl(π-C₃H₅)]₂–1 complex (5 mol % of Pd, Pd/1 = 1) gave **10** only in 50% yield along with the corresponding (*E*)-isomer (10%). For the coupling of organostannanes with aryl halides catalyzed by Pd–1, see: Shirakawa, E.; Yoshida, H.; Takaya, H. *Tetrahedron Lett.* **1997**, *38*, 3759–3762.